

## Material Safety Data Sheet

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### Section I - Chemical Product And Company Identification

**Product Name: HB Concrete Remover**

**Synonyms/Common Names**

Muriatic Acid, Hydrogen Chloride Solution, Chlorohydric Acid, HCl

CAS Number: N/A

HBCC MSDS No. CC26000



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### Section II - Composition/Information on Ingredients

Chemical Name	CAS Number	%	Exposure Limits (TWAs) in Air		
			ACGIH TLV	OSHA PEL	STEL
Hydrochloric Acid (Hydrogen Chloride)	7647-01-0	< 32%	2 ppm	5 ppm	N/A
Ethyl-ethoxy Propionate	763-69-9	< 2%	N/A	N/A	N/A

### Section III - Hazard Identification

**Routes of Exposure:** This product may affect the body either through ingestion, inhalation, or contact with the eyes and/or skin.

**Summary of Acute Health Hazards**

**Ingestion:** If ingested, solutions can cause corrosive burns to the mouth, throat, esophagus and stomach. Symptoms may include difficulty in swallowing, intense thirst, nausea, vomiting, diarrhea and in severe cases, collapse and death. Small amounts of acid which enter the lungs during ingestion or aspiration while vomiting can cause serious lung injury and death.

**Inhalation:** Vapor or mist from concentrated solutions can cause severe nasal irritation, sore throat, choking, coughing and difficulty breathing (50-100 ppm). Prolonged exposures can cause burns and ulcers to the nose and throat. Severe exposures (e.g. 1000-2000 ppm), for even a few minutes, can cause a life-threatening accumulation of fluid in the lungs (pulmonary edema). Symptoms of pulmonary edema such as shortness of breath can be delayed for several hours after the exposure.

**Skin:** Contact with the skin may cause severe irritation, skin burns and permanent skin damage. Prolonged exposure may result in ulcerating burns which could leave scars. Prolonged and repeated exposure to dilute solutions often causes irritation, redness, pain and drying and cracking of the skin.

**Eyes:** Contact with the eyes may cause severe irritation, eye burns and permanent eye damage, which may result in permanent blindness. Low concentrations of vapors or mist (10-35 ppm) can be immediately irritating, causing redness.

**Summary of Acute Health Hazards:** This solution is corrosive, and can burn and damage

eyes, skin, mucous membranes, and any other exposed tissue. If inhaled, irritation of the respiratory system may occur, with coughing, and breathing difficulty. Though unlikely to occur during occupational use, ingestion of large quantities may be fatal.

**Summary of Chronic Health Hazards:** Prolonged and repeated exposure to dilute solutions often causes irritation, redness, pain and drying and cracking of the skin. Repeated exposure to low concentrations of mist can cause brownish discoloration and damage to tooth enamel. Dental erosion becomes more severe with increased exposure. Repeated exposure to low concentrations can cause nose and gum bleeding. Chronic bronchitis and stomach pain (gastritis) have also been reported.

**Effects of Overexposure:** The most significant routes of occupational overexposure are inhalation and contact with skin and eyes.

**Medical Conditions Generally Aggravated by Exposure:** Hydrogen chloride (Hydrochloric Acid) is a respiratory irritant. Persons with impaired pulmonary function may be at increased risk from exposure. Periodic surveillance is indicated.

**Note to Physicians:** This product may cause severe pneumonitis if aspirated. If ingestion has occurred less than 2 hours earlier, carry out careful gastric lavage; use endotracheal cuff if available, to prevent aspiration. Observe patient for respiratory difficulty from aspiration pneumonitis. Give artificial resuscitation and appropriate chemotherapy if respiration is depressed. Following exposure the patient should be kept under medical review for at least 48 hours as delayed pneumonitis may occur. DO NOT attempt to neutralize the acid with weak bases since the reaction will produce heat that may extend the corrosive injury.

#### Section IV - First Aid Measures

**Ingestion:** If this product is swallowed, CALL PHYSICIAN OR POISON CONTROL CENTER FOR MOST CURRENT INFORMATION. If professional advice is not available, do not induce vomiting. Victim should rinse mouth with large amounts of water. Victim should drink one glass of water to dilute the ingested material. If spontaneous vomiting occurs, have victim lean forward with head down to avoid breathing in of vomitus, rinse mouth and administer more water. Never induce vomiting or give water to someone who is unconscious having convulsions, or who cannot swallow.

**Inhalation:** If vapors, mists, or sprays of this product are inhaled, remove victim to fresh air. Give artificial respiration ONLY if breathing has stopped. Do not use mouth-to-mouth method if victim ingested or inhaled the substance: induce artificial respiration with the aid of a pocket mask equipped with a one-way valve or other proper respiratory medical device. Give Cardiopulmonary Resuscitation (CPR) if there is no pulse AND no breathing. Obtain medical attention IMMEDIATELY. Symptoms may appear up to 48 hours after exposure.

**Skin:** Immediately flush contaminated skin with water and wash with soap and water. If large areas of the body are contaminated or if clothing is penetrated, immediately use safety shower preferably removing clothing while under the shower. Flush exposed areas with large amounts of water for at least 15 minutes. Keep affected area cool. GET PROMPT MEDICAL ATTENTION. Wash clothing before reuse. Destroy contaminated shoes.

**Eyes:** Immediately flush eyes with a directed stream of water for at least 15 minutes. Forcibly hold eyelids apart to ensure complete irrigation of all eye and lid tissue. Do not use chemical antidotes. Speed is essential. GET IMMEDIATE MEDICAL ATTENTION.

#### Section V - Fire Fighting Measures

**Flash Point:** None

**Autoignition Temperature:** N/A

**Lower Explosive Limit:** N/A

**Upper Explosive Limit:** N/A

**Unusual Fire and Explosion Hazards:** This product is corrosive, and presents a significant inhalation and contact hazard to fire-fighters. This product will not decompose at temperatures below 1500°C (2730°F). Thermal oxidative decomposition produces toxic

chlorine fumes and explosive hydrogen gas. Reacts with active metals (potassium, sodium, calcium, powdered aluminum, zinc, magnesium) to produce flammable hydrogen gas which can form explosive mixtures. May also form hydrogen chloride, and acid vapors. Explosive concentrations of hydrogen may accumulate inside metal equipment.

**Extinguishing Media:** Use water spray, fog, alcohol-resistant foam, dry chemicals, CO<sub>2</sub>, or other agents as appropriate for surrounding fire. Neutralize with soda ash or slaked lime. Do NOT use straight streams of water. Most foams will react with the material and release corrosive/toxic gases. Do not use carbon dioxide if cyanides are involved in a fire. Water fog is effective for controlling vapors. Controlled water addition is an effective method to reduce vapor pressure and control vapor emissions.

**Special Firefighting Procedures:** Use self-contained breathing apparatus and full protective equipment. If possible, prevent run-off water from entering storm drains, bodies of water, or other environmentally sensitive areas.

### Section VI - Accidental Release Measures

Spill and Leak Response: uncontrolled releases should be responded to by trained personnel using pre-planned procedures. Proper protective equipment should be used. In case of a spill, clear the affected area, protect people and respond with trained personnel. The proper personal protective equipment for incidental releases (e.g.-1 L of the product released in a well-ventilated area) use impermeable gloves, specific for the material handled, goggles, face shield, respirator and appropriate body protection.

In the event of a large release, wear proper protective equipment, use impermeable gloves, specific for the material handled, chemically resistant suit and boots, and hard hat. Self-Contained Breathing Apparatus or respirator may be required where engineering controls are not adequate or conditions for potential exposure exist. When respirators are required, Select NIOSH/MSHA approved based on actual or potential airborne concentrations in accordance with latest OSHA and/or ANSI recommendations.

Deny access to the area. Determine isolation distance. Stop leak at source, dike area, pick up with pump as much material as possible, prevent material from entering waterway, prevent contact with other chemicals. Absorb spilled liquid with polypads or other suitable absorbent materials. Neutralize residue with lime or soda ash or other acid-neutralizing agent. Decontaminate the area thoroughly. Test area with litmus paper to confirm neutralization. Place all spill residues in a suitable container. Dispose of in accordance with Federal, State and local hazardous waste disposal regulations (see Section XIII)

### Section VII - Handling and Storage

All employees who handle this material should be trained to handle it safely. Avoid breathing mists or sprays generated by this product. Use in a well-ventilated location.

**For Non-Bulk Containers** - Open containers slowly, on a stable surface. Containers of this product must be properly labeled. Only store in acid-resistant containers. Store containers in a cool, dry location, away from direct sunlight, sources of intense heat, or where freezing is possible. Material should be stored in secondary containers, or in a diked area, as appropriate. Store containers away from incompatible chemicals. Keep container tightly closed when not in use. Wash thoroughly after using this material. Storage areas should be made of fire-resistant materials. If appropriate, post warning signs in storage and use areas. Empty containers may contain residual liquid. Therefore, empty containers should be handled with care.

**Bulk Containers** - All tanks and pipelines which contain this material must be labeled. All equipment must be designed for use with this product. Perform routine maintenance on tanks or pipelines which contain this product. Report all leaks immediately to the proper personnel.

**Tank Car Shipments** – Tank cars carrying this product should be loaded and unloaded in strict accordance with tank-car manufacturer’s recommendation and all established on-site safety procedures. Appropriate personal protective must be used (see Section VIII). All loading and unloading equipment must be inspected prior to each use. Loading and unloading operations must be attended, at all times. Tank cars must be level, brakes must be set or wheels must be locked or blocked prior to loading and unloading. Tank car (for loading) or storage tank (for unloading) must be verified to be correct for receiving this product and properly prepared, prior to starting the transfer operations. All equipment must be designed for use with this product. Hoses must be verified to be clean and free of incompatible chemicals, prior to connection to the tank car or vessel. Valves and hoses must be verified to be in the correct positions, before starting transfer operations. A sample (if required) must be taken and verified (if required) prior to starting transfer operations. All lines must be blown-down and purged before disconnecting them from the tank car or vessel.

**Protective Practices During Maintenance of Contaminated Equipment** –Make certain application equipment is locked and tagged-out safely. Always use this product in areas where adequate ventilation is provided. Decontaminate equipment before maintenance begins by a triple-rinse with water followed, if necessary, by using an acid neutralizing agent and an additional rinse. Collect all rinsates and dispose of according to applicable Federal, State, or local regulations.

**Section VIII - Exposure Controls/Personal Protection**

**Respiratory Protection:** Use approved organic vapor acid-gas respirator for areas where airborne exposure is excessive. For a higher level of protection use positive pressure supplied air respiration protection or self-contained breathing apparatus or if oxygen levels are below 19.5% or are unknown.

**Ventilation:** Provide good general room ventilation to minimize exposure (typically 10 air changes per hour). Use local exhaust and corrosion-resistant ventilation at points of vapor emission. System should be discharged into absorption media.

**Protective Clothing:** Wear protective gloves such as rubber or neoprene to minimize skin contact. Use of rubberized coveralls and rubber shoes are suggested. Wash thoroughly after use. In case of emergency, or where there is a possibility of considerable exposure, wear complete acid suit with hood and forced air or self contained breathing apparatus.

**Eye Protection:** Wear safety glasses with side shields or chemical goggles. Person subject to hydrochloric acid exposure should not wear contact lenses. Face shields are recommended when the operation can generate splashes, sprays or mists. **Other**

**Protective Clothing or Equipment:** Eye wash and safety shower should be in close proximity.

**Work/Hygienic Practices:** Wash hands thoroughly with soap and water before eating, drinking, smoking or using toilet facilities. Do NOT place food, coffee or other drinks in the area where dusting or splashing of solutions is possible.

**Section IX - Physical and Chemical Properties**

<b>Physical State:</b> Liquid	<b>pH:</b> < 1.0
<b>Melting Point/Range:</b> -74°C; -101°F	<b>Boiling Point/Range:</b> --88.33°C; -127°F
<b>Appearance/Color/Odor:</b> Amber liquid with a sharp, pungent, irritating odor.	
<b>Solubility in Water:</b> Complete	<b>Vapor Pressure (mmHg):</b> 20 @ 20°C; 68°F
<b>Specific Gravity (Water=1):</b> 1.16-1.17	<b>Molecular Weight:</b> N/A
<b>Vapor Density (Air=1):</b> 1.27	<b>% Volatiles (by volume):</b> 98.4

**Density (lb/gal@15.6°C):** 9.68

**How to detect this compound:** Litmus paper will turn red upon contact with even low concentrations of this solution.

### Section X - Stability and Reactivity

**Stability:** Stable

**Hazardous Polymerization:** Will not occur

**Conditions to Avoid:** Heat or fire, runoff to sewer, inhalation of gas, sparks where hydrogen may be present.

**Materials to Avoid:** Contact with metals and strong oxidizers. Reacts exothermically with alkalis, metal oxides, amines, active metals carbonates, and sulfides. Reacts with oxidizers to give chlorine gas. Reacts with cyanides to give hydrogen cyanide gas. Reacts with sulfides to give hydrogen sulfide gas. Reacts with formaldehyde to give bischloromethyl ether (an OSHA regulated carcinogen). Reacts with amines to form ammonia. Reacts with carbonates to form carbon dioxide. Other materials to avoid are: Bases, acetic anhydride, alkali metals, aluminum, copper, copper alloys, fluorine, iron, sodium hydroxide, steel, sulfuric acid, vinyl acetate, zinc, potassium permanganate, cesium acetylene carbide, rubidium acetylene carbide, rubidium carbide, sodium, chlorosulfonic acid, oleum carbonates, perchloric acid, calcium phosphide, metal oxides, acetates, cesium carbide, beta-propiolactone, ethyleneimine, propylene oxide, lithium silicides, alcohols + hydrogen cyanide, 2-aminoethanol, ammonium hydroxide, calcium carbide, 1,1-difluoroethylene, ethylene diamine, magnesium boride, mercuric sulfate, silver perchlorate + carbon tetrachloride, uranium phosphide.

**Hazardous Decomposition Products:** Flammable hydrogen gas can be produced by the reaction with most metals. Chlorine will be released by mixing with strong oxidizers. Hydrogen chloride, carbon monoxide, carbon dioxide. When heated to decomposition, emits toxic hydrogen chloride fumes and will react with water or steam to produce heat and toxic and corrosive fumes. Thermal oxidative decomposition produces toxic chlorine fumes and explosive hydrogen gas.

### Section XI - Toxicological Information

LD50 (rabbit): 900 mg/kg. @ 100% HCl.

LD50 (rat): 3124 ppm/1 hour @ 100% HCl.

LC50 (inhalation, mouse) = 1108 ppm/1 hr.

### Section XII - Ecological Information

Rapidly hydrolyzes when exposed to water. Will exhibit extensive evaporation from soil surfaces. Upon transport through the soil, hydrochloric acid will dissolve some of the soil materials (especially those with carbonate bases) and the acid will neutralize to some degree.

LC50 mosquito fish = 282 mg/l 96 hours

LC50 fathead minnow = 21900 ug/l 96 hours

LC50 trout = 10 mg/l 24 hours

LC50 shrimp = 100 to 330 mg/l 48 hours (salt water)

LC50 gold fish = 178 mg/l 48 hours (salt water)

### Section XIII - Disposal Considerations

Dispose of spilled or wasted product, contaminated soil and other contaminated materials in licensed landfill or treatment facility in accordance with all local, state and federal regulations.

#### Section XIV - Transport Information

**DOT Proper Shipping Description:** UN1789, Hydrochloric Acid, 8, PG II

#### Section XV - Regulatory Information

**Reportable Quantity:** (CERCLA) 5000 Pounds (2270 Kilograms) (527.42 Gals)

**NFPA Rating:** Health - 3; Flammability - 0; Instability - 1

0=Insignificant 1=Slight 2=Moderate 3=High 4=Extreme

**Uniform Fire Code Rating:** Class 3 Water-Reactive Material

**Carcinogenicity Lists:** No **NTP:** No **OSHA Regulated:** No

**IARC Monograph:** Hydrogen Chloride - Group 3 Carcinogen

EPCRA Section 302 (EHS) TPQ: 500 Lbs.

EPCRA Section 304 (EHS) RQ: 5,000 Lbs.

CAA 112(r) TQ: 5,000 Lbs.

**Section 313 Supplier Notification:** This product contains the following toxic chemical(s) subject to the reporting requirements of SARA TITLE III Section 313 of the Emergency Planning and Community Right-To Know Act of 1986 and of 40 CFR 372:

<u>CAS #</u>	<u>Chemical Name</u>	<u>% By Weight</u>
7647-01-0	Inhibited Hydrochloric Acid (Hydrogen Chloride)	31.45%

#### Section XVI - Other Information

Inhibited Hydrochloric Acid contains approx. 30-35% Hydrogen Chloride

**Synonyms/Common Names:** Inhibited Muriatic Acid

**Chemical Family/Type:** Inorganic Acid

**Sections changed since last revision:** IV, VI, VII, VIII, IX

**IMPORTANT!** Read this MSDS before use or disposal of this product. Pass along the information to employees and any other persons who could be exposed to the product to be sure that they are aware of the information before use or other exposure. This MSDS has been prepared according to the OSHA Hazard Communication Standard [29 CFR 1910.1200]. The MSDS information is based on sources believed to be reliable. However, since data, safety standards, and government regulations are subject to change and the conditions of handling and use, or misuse are beyond our control, **Hill Brothers Chemical Company** makes no warranty, either expressed or implied, with respect to the completeness or continuing accuracy of the information contained herein and disclaims all liability for reliance thereon. Also, additional information may be necessary or helpful for specific conditions and circumstances of use. It is the user's responsibility to determine the suitability of this product and to evaluate risks prior to use, and then to exercise appropriate precautions for protection of employees and others.